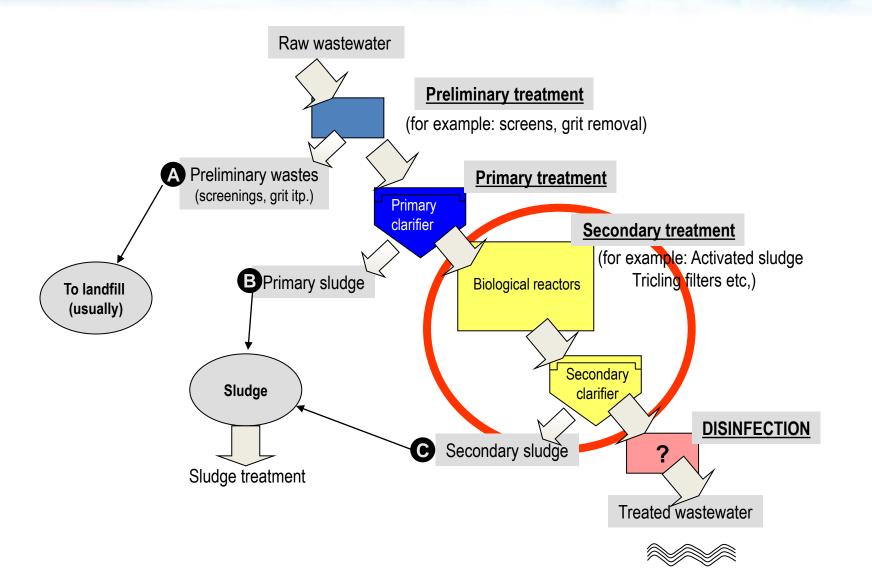


Wastewater Treatment Technology Lecture 6

Denitrification

WWTP overall scheme



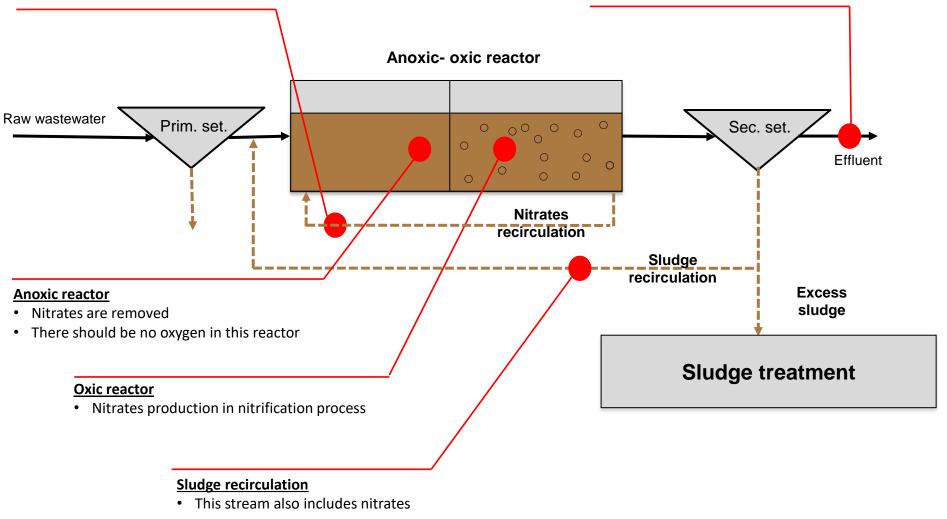
Simplest scheme

Nitrate recirculation

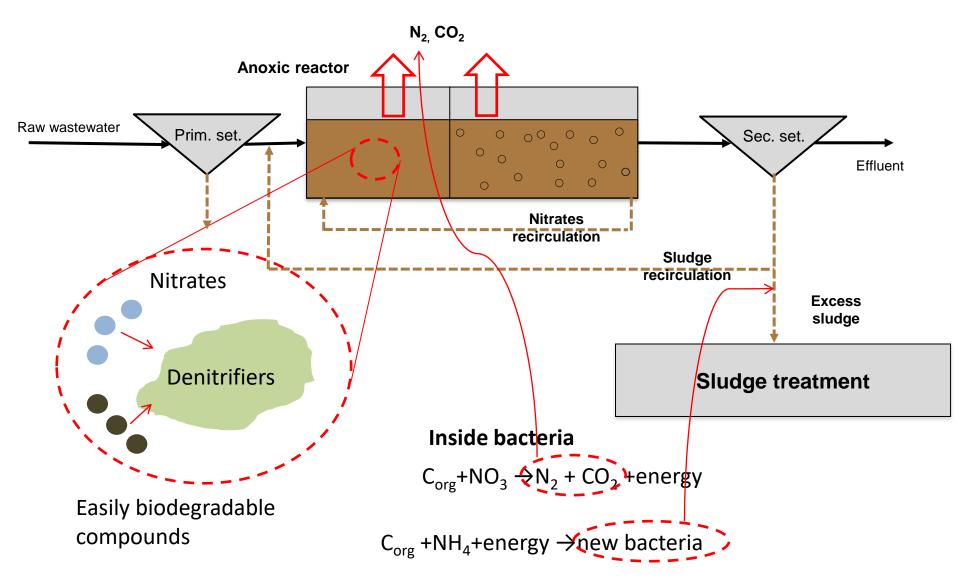
- Nitrites created in oxic reactor are recirculated
- Stream is intense usually few hundred % of Q_{inf}

Treated wastewater

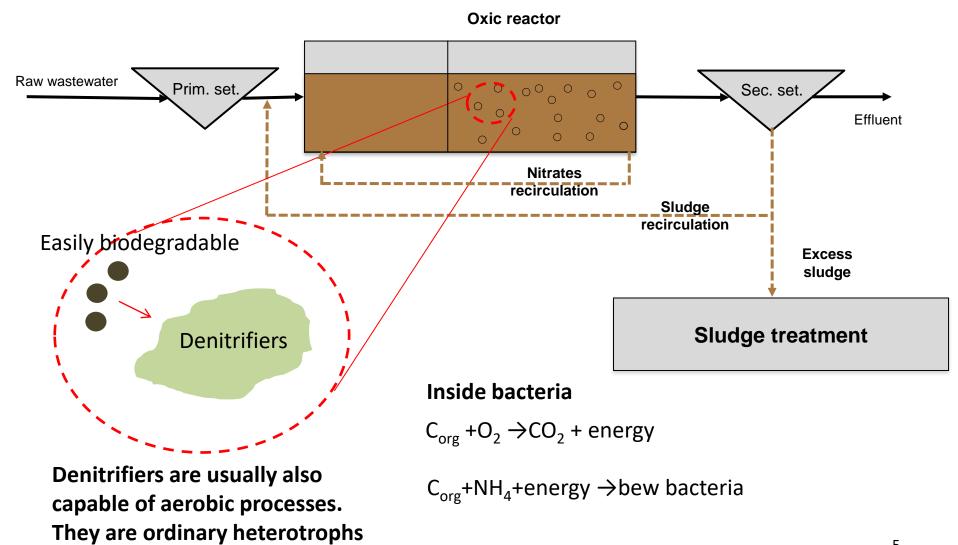
• Definetely less nitrates than in aerobic only process



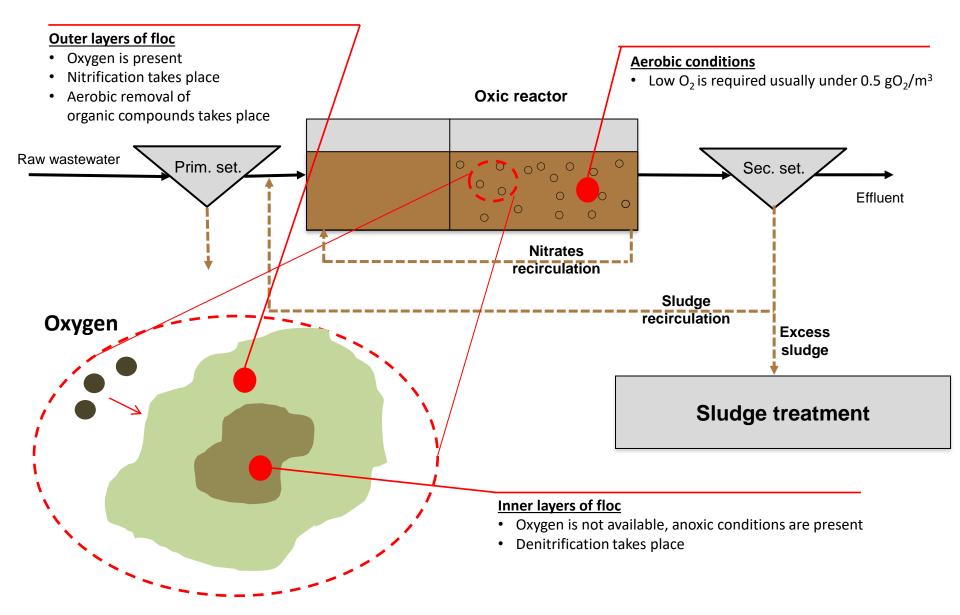
Course of denitrification



Denitrifiers in oxic reactor

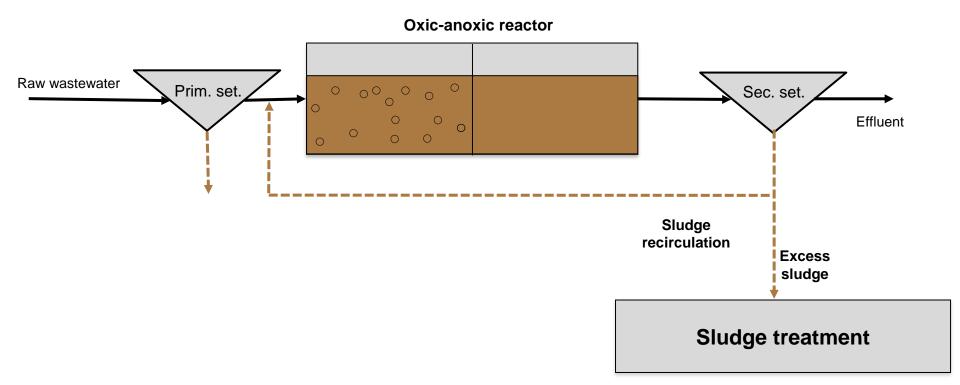


Denitrification in oxic reactor

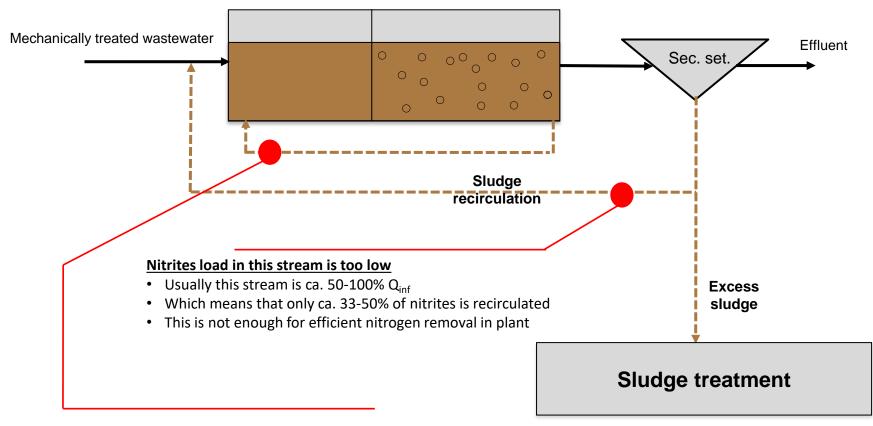


Activated sludge floc

Why not this scheme?



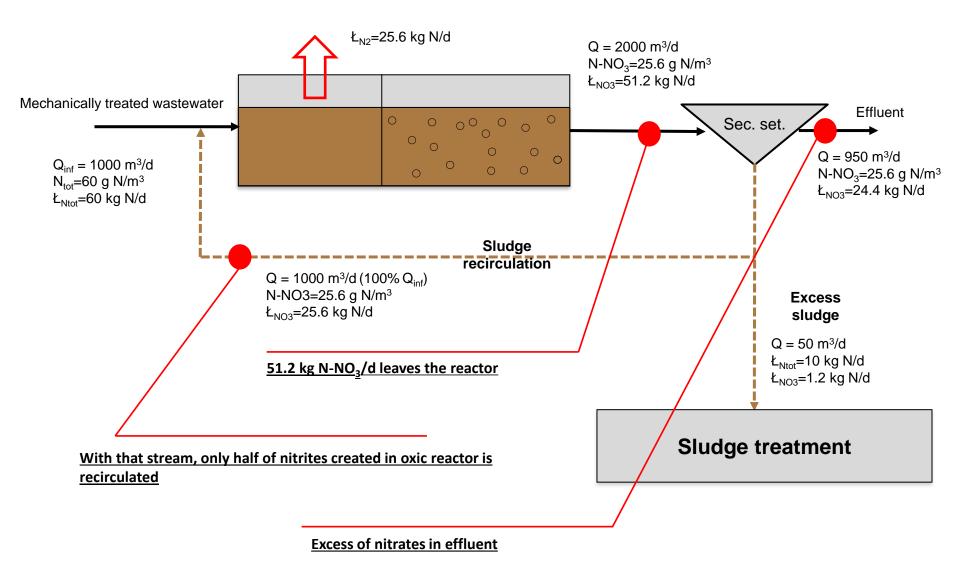
Role of nitrates recirculation



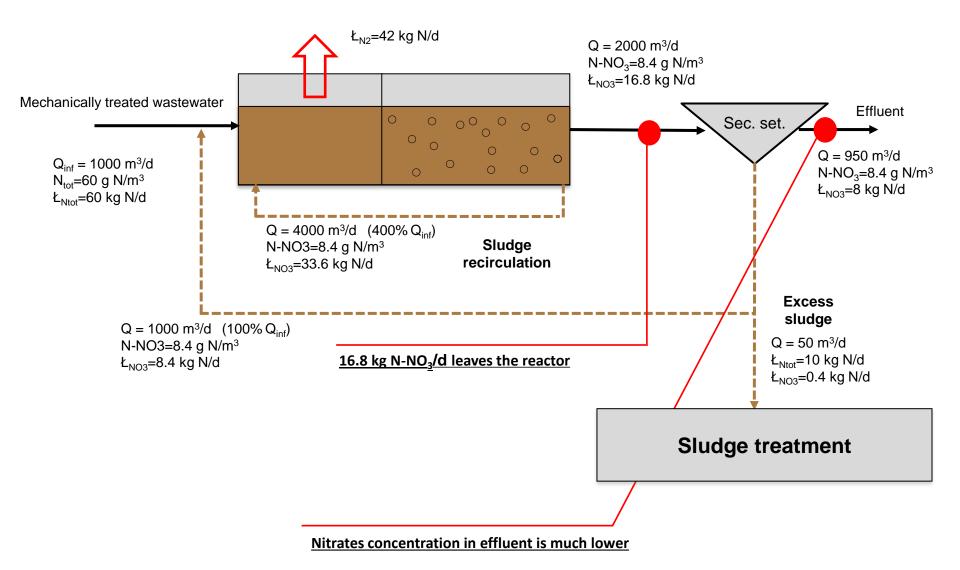
Aim of this recicrulation is to provide enough nitrates in anoxic reactor

- Usually this stream is few hundred % Q_{inf}
- This stream recirculates most of nitrates to anoxic reactor
- Thanks to that stream the required removal rate of nitrogen can be achieved

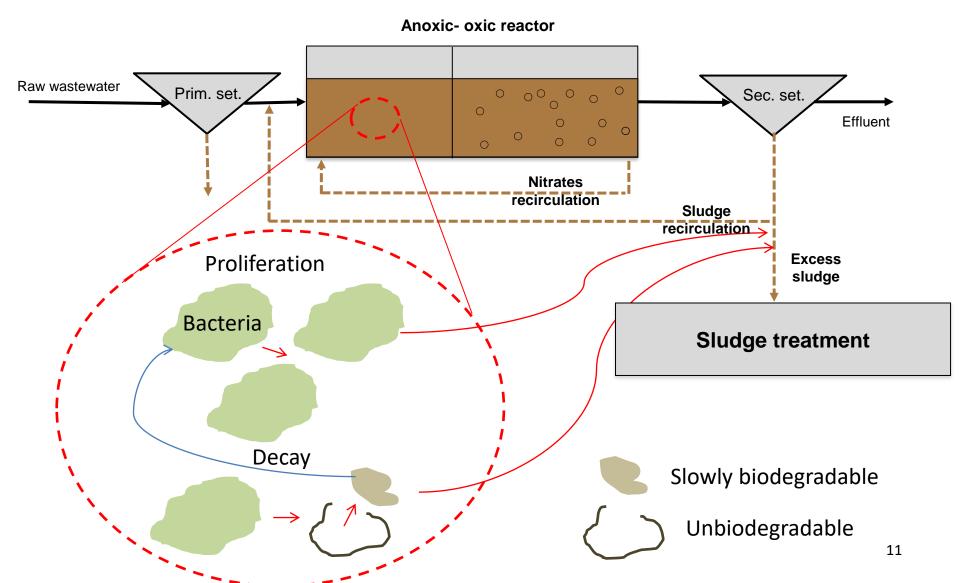
Role of nitrates recirculation



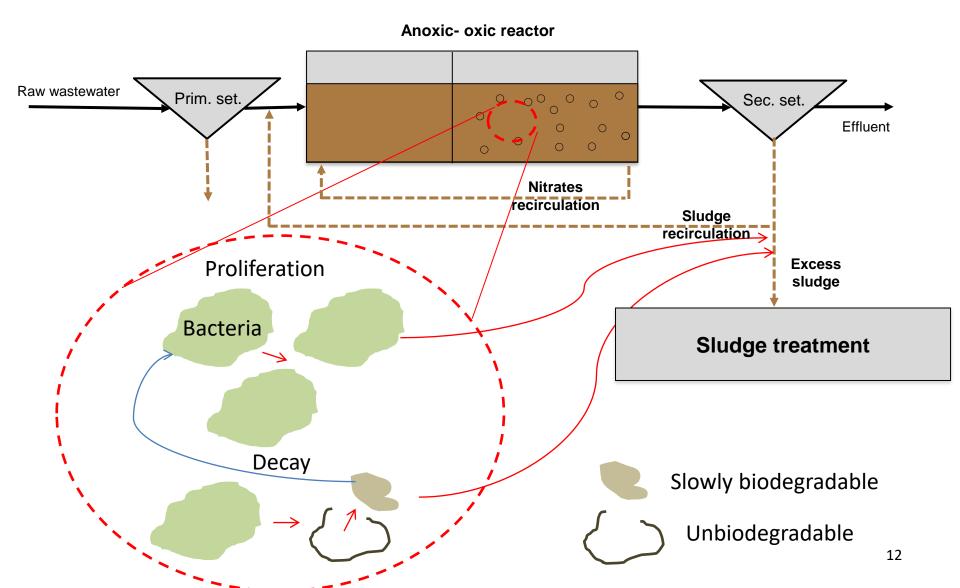
Role of nitrates recirculation



Denitrifiers



Denitrifiers



Basic data

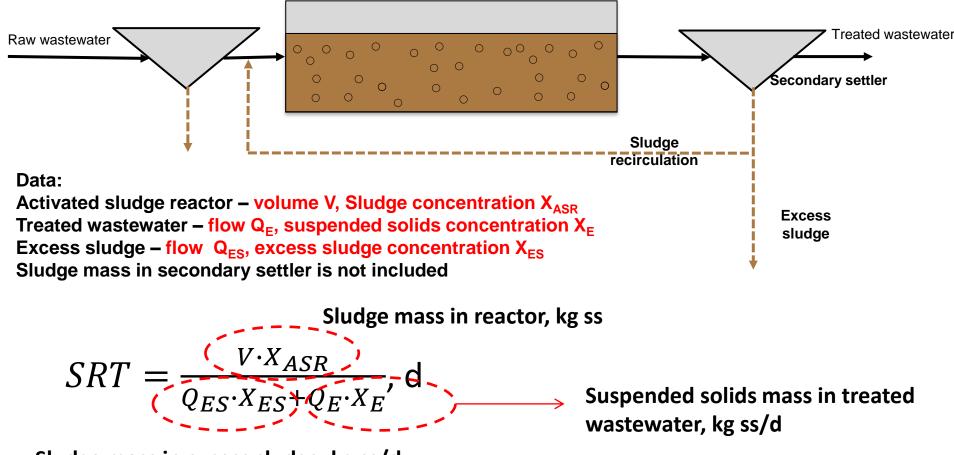
SRT> 10 d (in AX/OX) pH - 6.0 - 8.0 T >8°C $O_2 < 0.5 \text{ gO}_2/\text{m}^3$ in denitrification reactor



Aerobic sludge retention time

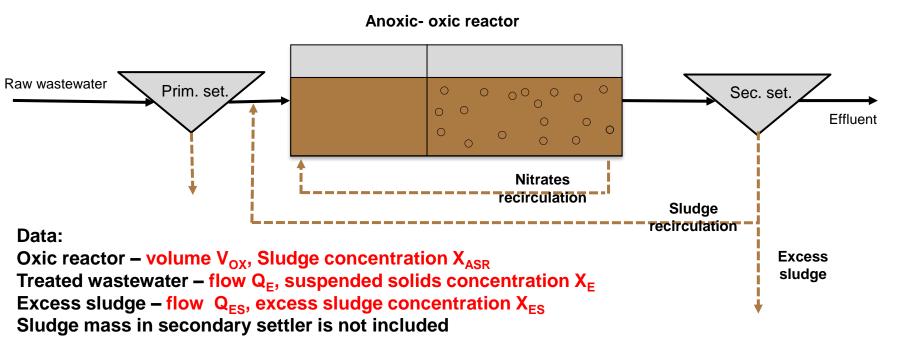
Sludge retention time definition (SRT)





Sludge mass in excess sludge, kg ss/d

Aerobic sludge retention time definition (SRT_{ox})

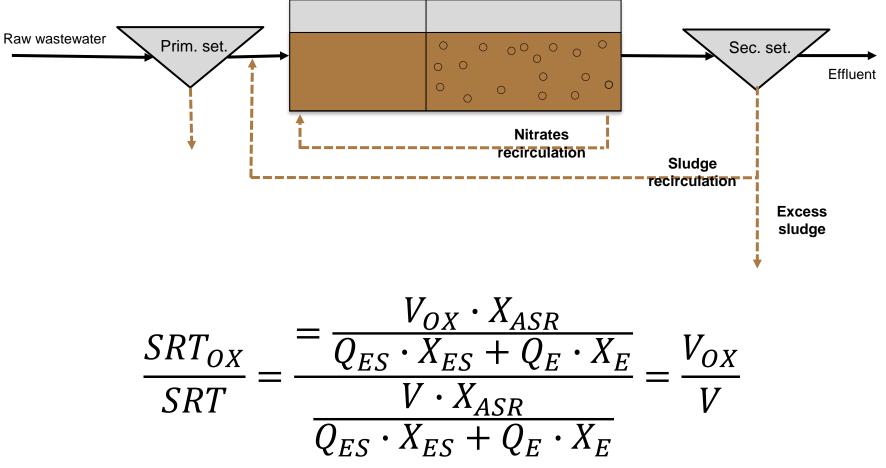


Sludge mass in oxic reactor, kg ss

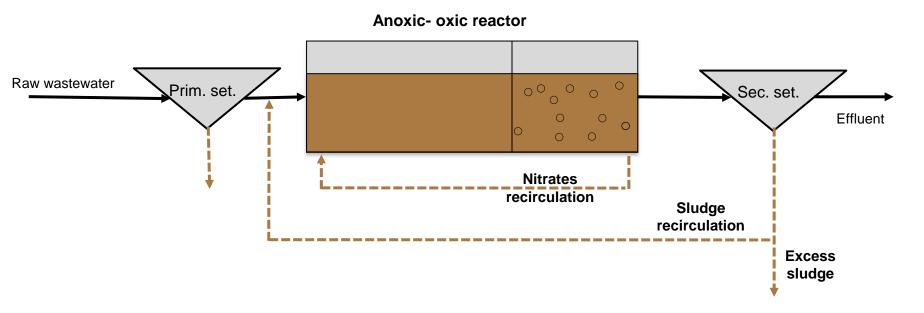
$$SRT_{OX} = \frac{V_{OX} \cdot X_{ASR}}{Q_{ES} \cdot X_{ES} + Q_E \cdot X_E}$$
, d $SRT = \frac{V \cdot X_{ASR}}{Q_{ES} \cdot X_{ES} + Q_E \cdot X_E}$, d

Aerobic sludge retention time definition (SRT_{ox})

Anoxic- oxic reactor



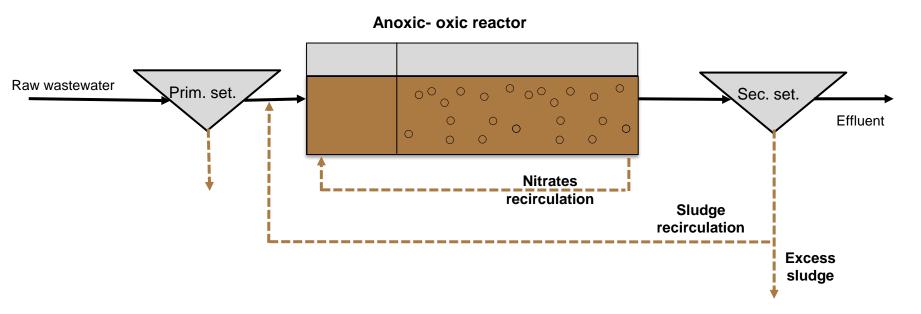
Warm period



Warm months:

- 1. Higher temperature
- 2. High nitrifiers growth rate
- 3. Aerobic sludge retention time can be lower
- 4. Oxic reactor's volume can be lower
- 5. Better nitrogen removal due to more efficient denitrification

Cold period



Cold months:

- 1. Lower temperature
- 2. Low nitrifiers growth rate
- 3. Aerobic sludge retention time has to be higher
- 4. Oxic reactor's volume has to be higher
- 5. Worse nitrogen removal due to less efficient denitrification

Effectiveness of nitrogen compounds removal

In good conditions: >95% NH₄ >90% NO₃ Ok. 90 % Ntot



- 1. Describe the denitrification process?
- 2. Under what conditions is denitrification possible in an oxygen chamber?
- 3. What are the substrates of the denitrification process?
- 4. What role does nitrate recirculation play?
- 5. What does nitrate removal look like when there is no nitrate recirculation?
- 6. What is the aerobic sludge retention time?
- 7. Operation of the AX system in summer and winter differences